DEVELOPMENT OF AN ULTRA-SAFE RECHARGEABLE LITHIUM-ION BATTERY



Contract # N00014-94-C-0141 ARPA Order # 9332004arp01/13 APR 1994/313ES

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Reporting Period: 16 June to 15 July, 1995

Submitted by:

The Electrofuel Manufacturing Company Inc.

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ARPA Order No.: 9332004arp01/13APR1994/313ES

Program Code No.: ARPA-BAA93-32

Contractor: The Electrofuel Manufacturing Company Inc. Contract No.: N00014-94-C-0141 Contract Amount: \$1271728

Effective Date of Contract: August 15, 1994 Expiration Date of Contract: February 14, 1996

Principal Investigator: J.K. Jacobs

Telephone No.: (800) 388-2865

Short Title of Work: Lithium-ion Battery Development

Reporting Period: June 16, 1995 to July 15, 1995

Description of Progress:

Work is accelerating in a number of areas simultaneously.

The major effort during this period was to start operating and debugging the assembled (flexible prototype production line) system. Upon assembly, the system did not work satisfactorily. The first approach was to use a plastic carrier material through the rollers. This was used to generally align the system. After the plastic material was changed to the Aluminum foil, 12 micron (0.5 thou), it was found that precision alignment of the system was necessary. An optical alignment system using a laser technique was developed using a He/Ne laser along with a number of flat mirrors. Two mirrors were attached to the base of the system and other mirrors were attached to the moving/rotating part of the system. This alignment allowed us to roll the Al foils without warping. This alignment was tested by rolling about 600 feet of Al foils.

Electrode coating: Two methods of coating electrode on the Al foils were developed, constructed and tested. This work was started in the last two months but was completed in this time The first approach was to feed the electrode material through a curtain coater slot under hydrostatic pressure. method has some advantages but because of the very strong dependence of viscosity of materials to the flow rate of the material, this method was difficult to control. The flow properties of To improve the flow properties this material is non-Newtonian. of this material (manganese oxide, Kynar, propylene carbonate), attempts were made to introduce a low amplitude vibration. low frequency (60 Hz) and high frequency (ultrasonic - 20,000 Hz) vibrations were introduced during the flow. This improved the flow uniformity and can be developed in the future. In general the cathode coating mix is observed to be very thixotropic with

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viscosity changes of several hundred times with increasing shear rate. This is generally characteristic of colloids near the iso electric point and is a difficult material to be used for curtain coatings.

But another method was also developed in parallel in order to provide a fast proof of concept for the flexible production line. This consisted of a modification of tape casting procedure implemented on-line in the lamination process. This method was devloped, constructed and tested. The positive electrode material was deposited on the aluminum foil with good uniformity of about 7.5% and an average thickness of 300 microns, before it was squeezed into 170 microns, which is the design electrode thickness.

The thickness of the copper foil used for the negative electrode was reduced from 25 microns to 18 microns. The adhesion of the copper foil to the negative electrode material is sufficient but not excellent. An electrochemical etching process is used to improve the copper foil adhesion characteristics.

Hand crafted cells continue to be produced and tested. Any changes in chemistry is implemented and tested through handcrafted cells. One of the major problem area is the interface adhesion in the cells. A major development this month on the chemistry was the development of an improved method for adhesion of separator to the electrodes. This was done using adhesives which also conduct lithium ions. Cells with these binders are being produced. A batch of six "glued cell" are being constructed and will be cycled soon.

For thin cells it is possible to get high power out of these Planar Lithium ion batteries. To test for high rate capability, unoptimised cells which were undergoing long term testing were used after the 325th cycle. Even operating with these unoptimised cells, it was found that excellent rate performance could be achieved. Discharge of these cells were carriedout at the 1C, 2C and 4C rate. The cells behaved well, and after the high rate discharges, they went back to the normal charge/discharge regime without showing any effect. It appears that thin film lithium ion cells could be a candidate for high rate applications, which are presently being filled by silver/zinc batteries.

A paper was given in the Myrtle Beach Conference outlining the performance and progress of this battery development program.

Change in Key Personnel: None

Summary of Substantive Information Derived from Special Events: None:

Problems Encountered and/or Anticipated: None

Action Required by the Government: None

	Total Est- imate of Program	Funding	Electro fuel Contri- bution
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